

IV.B. ELECTROLYTES & THEIR DYSFUNCTION

B.a. GENERAL

B.a.A. FUNCTIONS OF ELECTROLYTES

Table IVB-1. Functions of Electrolytes in the Body

IVB-1

(M)ajor/(L)esser Role in/as	Na ⁺	K ⁺	Ca ⁺⁺	PO ₄ ⁻³	H ⁺	HCO ₃ ⁻	NH ₄ ⁺	Mg ⁺⁺	Cl ⁻	Proteins
1a. Regulating body fluids	M ^a	L	L	L	L	L	L	L	L	M
b. Maintaining intracellular osmolality		M								
c. Determining normal intracellular K ⁺ levels								M		
d. Acid-base regulation	M	L		M ^b	M	M	L		M ^c	
2a. Coenzyme/cofactor		L	M	M ^d				M ^e		
b. Anabolic metabolism of carbohydrates & proteins		M								
c. Energy for anabolic processes, membrane transport, fibril contraction				M ^f						
3a. Major component of macromolecules & organelles (unit membrane & nucleic acids)				M						
b. Numerous vital chemical reactions	M									
4a. Second messengers			M ^{g,h}							
b. Controlling synaptic release of ACH			M							
c. Nerve/muscle impulse	M	M								
d. Initiation/inhibition of nerve/muscle impulse in post synaptic cells	M	M	L							
e. Maintains refractability of cardiac muscle			M							
5a. Blood clotting			M							
b. Normal function of white blood cells & platelets				M						
c. Component of HCl in the stomach					M				M	
d. Component of alkaline digestive juices						M				
e. Absorption of vitamin B ₁₂ & intrinsic factor in small & large intestine			M							
6. Component of bone matrix & otoliths			M	M		M ⁱ		L		

a major determinant of extracellular volume

b major buffer in blood; buffer in renal excretion of H⁺

c chloride shift

d essential in enzymes for metabolism of glucose, fat, & protein

e essential in all enzyme systems known to be catalyzed by **ATP**; replication, transcription & translation; fibril contraction & glycogen metabolism

f **ATP + 2,3 DPG**

g best example: Ca⁺⁺ binds to troponin on troponin-tropomyosin complex, → altered position of tropomyosin, → exposure of active sites on actin, → attraction of the head of the myosin filament, → the power stroke.

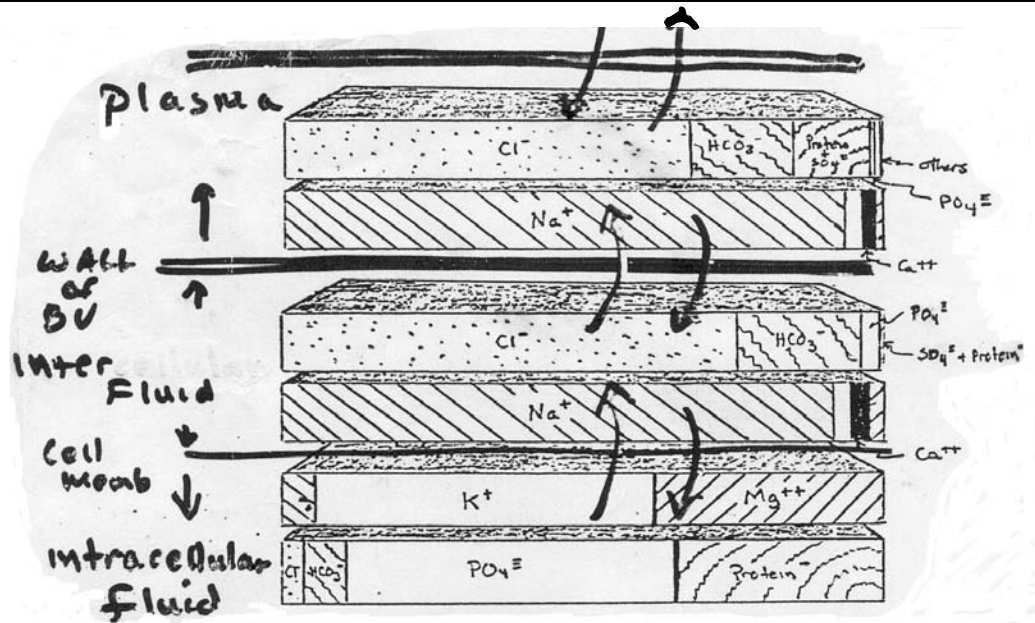
h K⁺ & Mg⁺⁺ can bind to Ca⁺⁺ receptors

i as CO₃⁼

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B.a.B. TERMINOLOGY

1) hypo-	3) -natr-		1)	3)
2) hyper-	4) -kal-		2)	4)
	5) -calcem-	8) -emia		5)
	6) -phosphat-			6)
	7) -magnes-			7)
				8)



IVB-2
Critical Figure IVB-1. Relative Concentrations of Electrolytes in Three Fluid Compartments

B.b. SODIUM

B.b.A. NORMAL GAINS & LOSSES (MDR = 300 mg/day) (normal intake: 12-30 times this)

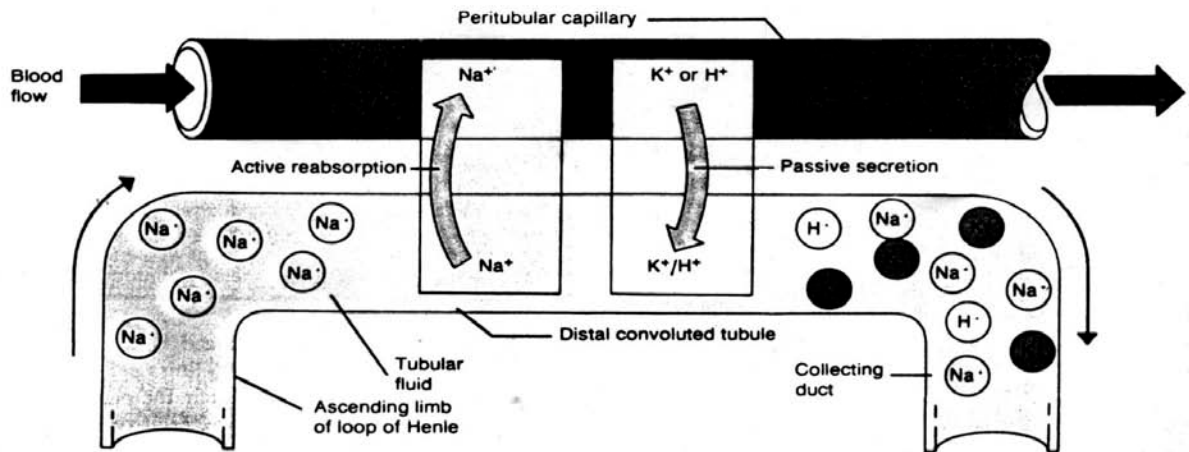
GAINS LOSSES
Diet kidneys (90%)

KIDNEY ACTIONS
↑ Na⁺ → ↑ Na⁺ excretion
↓ Na⁺ → ↑ Na⁺ retention

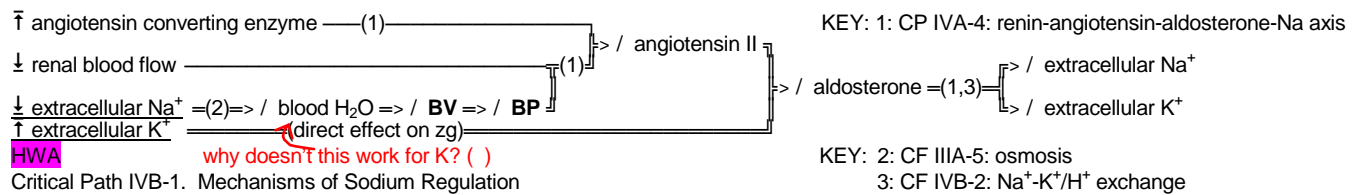
Meds skin (≤ 15-30 mg/day, except during profuse sweating)
gi tract (normally actively absorbed in lower bowel, so that in stools: 32 mEq Na⁺/ℓ)

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B.b.B. RENAL RETENTION [normal serum = 136-145 mmol/ℓ (200, p 676) = < > mEq/ℓ]



IVB-3
Critical Figure IVB-2. The Coupling of K⁺/H⁺ Secretion to Na⁺ reabsorption (22, p 767)



B.b.C. **HYPONATREMIA** [serum Na⁺ < 135 mEq/l < 120 mEq/l is serious (200, p 676)]

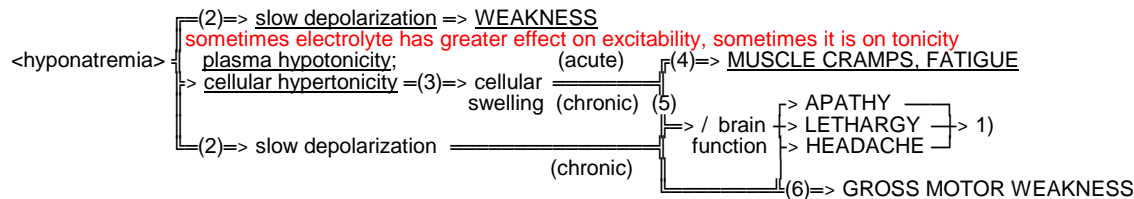
B.b.C.a. **CONDITIONS**

- 1) normal vomiting &/or diarrhea ⇒ (iso-osmotic fluid loss) ⇒ no Δ Na⁺
- 2) ↑ Na⁺ secretion in kidney
- 3) ↑ Na⁺ loss through skin
- 4) dilution of existing serum Na⁺ via other osmotically active substances (↑ sugar → ↑ H₂O → ↓ Na⁺)
- 5) ↓ Na⁺ gain w/ ↑↑ H₂O gain
- 6) ↑↑ Na⁺ loss w/ ↑ H₂O loss
- 7) ↑ Na⁺ loss to GI tract

B.b.C.b. **ETIOLOGIC FACTORS**

- kidney: ↑↑ diuretics, salt wasting kidney disease, Addison's Disease →)
- skin: 3rd degree burns →); heat exhaustion →)
- endocrine: diabetes →)
- GI tract: water intoxication →); frequent tap water enemas, frequent GI irrigation w distilled H₂O →)

water intoxication -(1)-> acute hyponatremia -> / Na⁺ filtration -> / urine osmolality -> / URINE SPECIFIC GRAVITY*



- 1) DISORIENTATION (110 mOsm/kg)
- CONFUSION → SEIZURES → COMA

- KEY: 1: P IVA-5: WVE path
 2: Table IVB-1: 4c: role of Na⁺ in action potential
 3: CF IIIA-5: osmosis & H₂O → hypertonic soln
 4: CP IIIA-2: mild hypoxia-> / ATP → / sarcoplasmic Ca⁺⁺
 5: CP IIIA-9: cellular swelling → / mitochondrial function
 6: CF IIIA-3: role a of ATP in the cell: microfibril contraction

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Path IVB-2. Manifestations of Hyponatremia

B.b.D. **HYPERNATREMIA** [serum Na⁺ > 148 mEq/l: > 160 mEq/l is serious (200, p 676)]

B.b.D.a. **CONDITIONS**

- 1) rapid, hypertonic Na⁺ ingestion/input
- 2) defect in thirst mechanism
- 3) inability to drink/get water
- 4) H₂O input < H₂O output
- 5) ↓ Na output

B.b.D.b. **ETIOLOGIC FACTORS**

- taking excess salt tablets, near drowning in salt water, rapid NaHCO₃ administration during CPR, osmotically active (hypertonic) tube feeding w/out adequate H₂O, therapeutic abortion w saline solution placed intravenously rather than intraamniotically →)
- head trauma →)
- oral trauma, severe laryngopharyngitis, unconsciousness, bed ridden, infancy →)
- fever, watery diarrhea, tracheobronchitis, diuretic therapy, strenuous exercise in heat, diabetes insipidus, →)

<congestive heart failure>, most forms of kidney disease, <Cushing's syndrome>* →)

B.c.. **POTASSIUM**

B.c.A. **NORMAL GAINS & LOSSES** (MDR = 50-100 mg/day) (↑ demands during trauma & stress)

GAINS	STORAGE	LOSSES
diet	cells	kidneys (80-90%), skin, stools

B.c.B. **REGULATION** (normal serum = 3.5-5.0 mEq/l) (K^+ in plasma = 2% of total body K^+)

B.c.B.a. **RENAL MECHANISMS** (review: (1) CP IVB-1: r,a,al,all,a axis & (2) CF IVB-2: action of aldosterone on dct)

- 1) **aldosterone** (1,2): (\uparrow in serum $K^+ \geq 1 \text{ mEq/l} \rightarrow 3X$ / in aldosterone)
- 2) DCT (2): $\bar{}$ serum $K^+ - (2) \rightarrow / K^+$ secretion \rightarrow / H^+ secretion \rightarrow / H^+ retention \rightarrow metabolic acidosis
- 3) DCT (2): \downarrow serum $K^+ - (2) \rightarrow / K^+$ secretion \rightarrow / H^+ secretion \rightarrow / H^+ retention \rightarrow metabolic alkalosis
- 4) metabolic alkalosis $- (3) \rightarrow /$ plasma $HCO_3^- \rightarrow / KHCO_3$ secretion $\rightarrow //$ serum K^+ levels [(3) S. IV.C.c.c.]

B.c.B.b. **REDISTRIBUTION BETWEEN FLUID COMPARTMENTS** (review: (4): CF IIIA-5: cell lysis)

- 1) **insulin** (\uparrow serum $K^+ \geq 1 \text{ mEq/l} \rightarrow /$ serum insulin \rightarrow / K^+ into cells)
- 2) injury to cells $- (4) \rightarrow /$ serum K^+
- 3) cellular anabolism > cellular catabolism (K^+ necessary for glycogen synthesis & storage, & protein synthesis) $\rightarrow \downarrow$ serum K^+ & *visa versa* (Table IVB-1, 2b)
- 4) K^+/H^+ exchange (cells vs secretion into DCT)
 - $\uparrow K^+$ into cells, $\uparrow H^+$ out of cells & $\uparrow K^+$ out of cells, $\uparrow H^+$ into cells
 - acidosis \rightarrow / H^+ into cells \rightarrow / K^+ out of cells (1:1 ratio); alkalosis, just the reverse
- 5) **epinephrine** (\uparrow epinephrine $\rightarrow \uparrow K^+$ into cells)
- 6) $\downarrow Mg^{++} - ?? \rightarrow \downarrow$ intracellular K^+ (Table IVB-1, 1c)

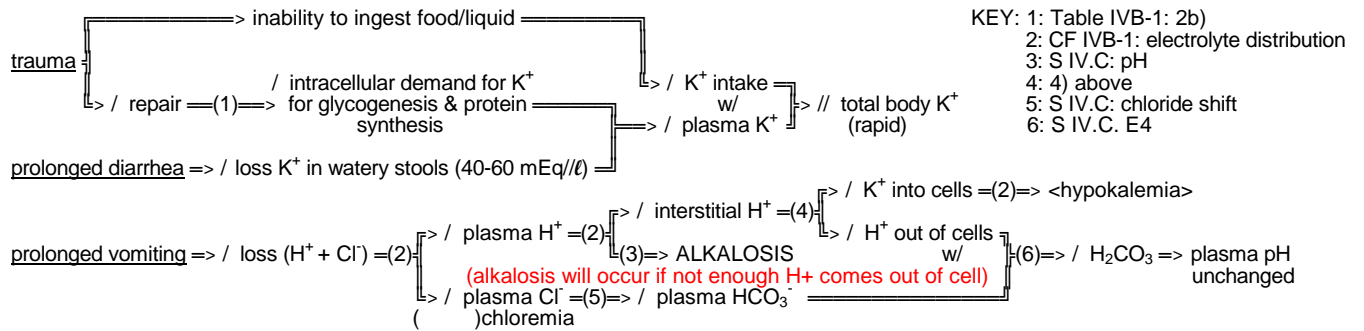
B.c.C. **HYPOKALEMIA** [serum $K^+ < 3.5 \text{ mEq/l}$]

B.c.C.a. **ETIOLOGIC FACTORS**

- 1) redistribution between intracellular & extracellular compartments
- 2) inadequate intake
- 3) excessive intestinal losses
- 4) excessive renal losses

B.c.C.b. **PATHOPHYSIOLOGY**

B.c.C.b.(A.) **ETIOLOGY OF HYPOKALEMIA** NB: kidneys have no mechanism for conserving K^+ during acute K^+ loss, ...)



Conn's syndrome $-T \rightarrow \downarrow$ total body $K -$ (slow)

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Path IVB-3. General Etiology of Hypokalemia

B.c.D.b.(B.) **MANIFESTATIONS OF HYPERKALEMIA**

- KEY: 1: CF IVB-1: electrolyte distribution
 2: F IIIA-6: electrochemical gradient & cell
 3: P IVB-4: effect of \uparrow/\downarrow intracellular K^+ on resting potential

$<\bar{r} K^+> = (1) \Rightarrow$ / K^+ / flow outside $= (2) \Rightarrow$ of $K^+ = (3) \Rightarrow$ (polarization) \Rightarrow / excitability/irritability

(201, p 106)

1) \Rightarrow resting potential above threshold potential \Rightarrow **CARDIAC ARREST**

(6 mEq//l)
 PARESTHESIA
 (G sensation)
 (3) $\left\{ \begin{array}{l} \text{ARRHYTHMIAS*} \\ \text{CHANGED ECG*} \end{array} \right\} T=(if: \bar{r} \bar{r} K^+ \Rightarrow 1)$
 (F 16.15, p 417)

IVB-6

Path IVB-6. General Manifestations of Hyperkalemia

B.d. **CALCIUM & PHOSPHATE BALANCE**

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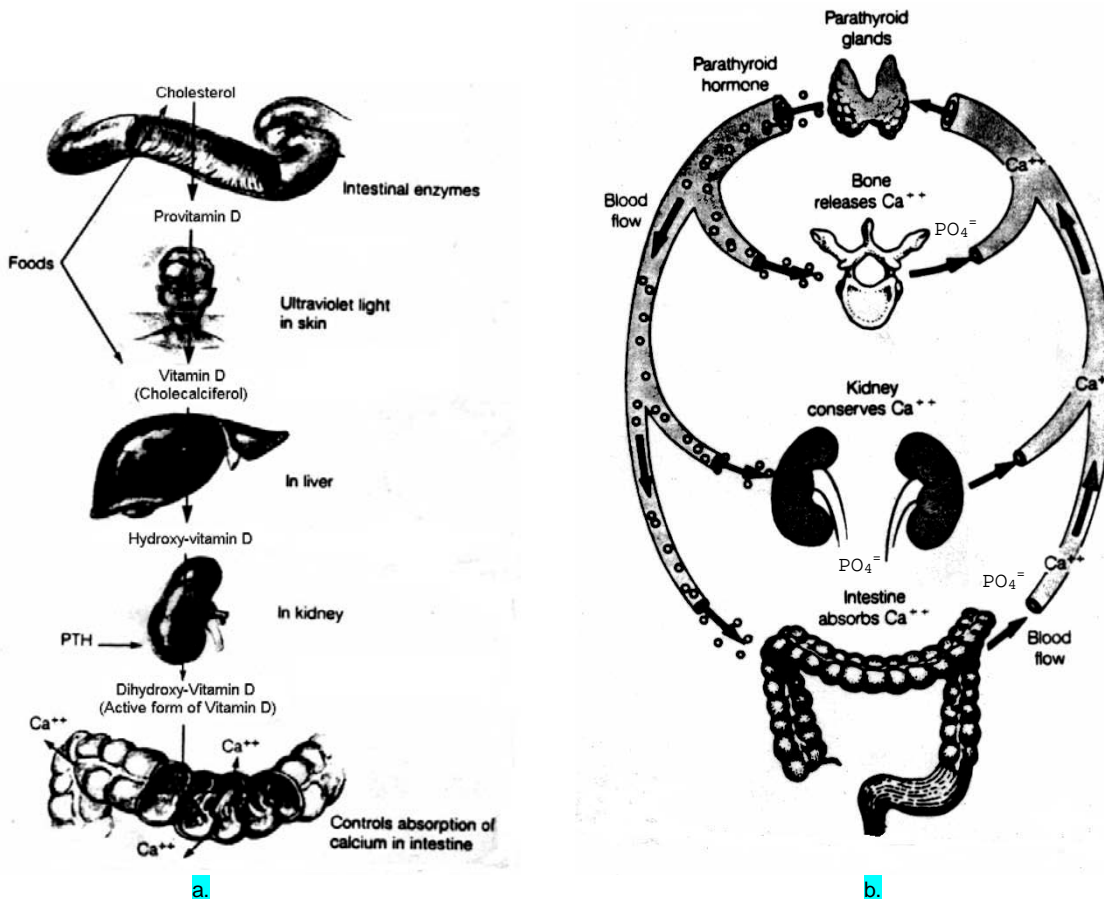
B.d.A. **BACKGROUND**

- 1) Ca^{++} & PO_4^{-3} are linked because main reservoir is bone which has 70% crystalline salts, most of which are $Ca_3(PO_4)_2$
- 2) in adults, normal serum $Ca^{++} = 8.8-10.0$ mg/dl (4.4-5.0 mEq/l); normal serum $PO_4^{-3} = 2.7-4.5$ mg/dl (0.85-1.4 mEq/l)
- 3) concentration in extracellular fluids is regulated such that their products in mg/dl = 35 (to prevent precipitation of $Ca_3(PO_4)_2$ into tissues)

B.d.B. **NORMAL GAINS, STORAGE & LOSSES**

<p>GAINS gastrointestinal^a</p>	<p>STORAGE bones</p>	<p>LOSSES^a kidneys feces</p>	<p>^a30-50% of Ca^{++} ingested is absorbed in duodenum & upper jejunum; remainder eliminated through bowels; <u>all</u> of PO_4^{-3} ingested is absorbed.</p>
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B.d.C. **REGULATION**



IVB-6

Figure IVB-3. Roles of Vitamin D & PTH in Ca^{++} & PO_4^{-3} regulation (22, p 482)

↓ serum Ca⁺⁺ below critical level } / PTH (1) => / Ca⁺⁺ excretion => / serum Ca⁺⁺ (1) => / PO₄⁻³ excretion => / serum PO₄⁻³ =? ()

Key: 1: F IVB-3: role of PTH on kidney

IVB-7a

Path IVB-7. Normal Mechanisms for Calcium & Phosphate Regulation

B.e. **CALCIUM** [normal serum = 8.8 - 10 mg/dl]

B.e.A. **EXTRACELLULAR Ca⁺⁺ LEVELS** (99% in bones & teeth, 1% in tissue cells & extracellular fluid)

ef => ↓ serum Ca⁺⁺ below normal levels (nl) => / PTH => / activity of osteoclasts => / Ca⁺⁺ erosion from bone => / serum Ca⁺⁺ to nl

ef => ↑ serum Ca⁺⁺ above nl => / calcitonin => / activity of osteoblasts => / Ca⁺⁺ deposition to bones => / serum Ca⁺⁺ to nl

IVB-7b

Path IVB-8. Mechanisms of Calcium Regulation

Key: 1: P IVB-7: response of parathyroid gland to low calcium

of total serum Ca⁺⁺

1) 40% bound to plasma proteins ∴ can't leave plasma

36% bound to albumin; 4% bound to globulin ∴ Δ albumin → Δ total serum Ca⁺⁺

(total serum Ca⁺⁺ ↓ 's 0.75-1 mg/dl for every 1 g/dl ↓ in serum albumin)

↑ pH → ↑ binding of Ca⁺⁺ to plasma protein (serum Ca⁺⁺ ↓ 's 0.16 mg/dl for every 0.1 ↑ in pH)

2) 10% bound to citrates, phosphates & sulfates (citrates used to inhibit coagulation in blood transfusions;

cleared by liver in minutes; citrates → ↑ Ca⁺⁺ binding → ↓ free Ca⁺⁺ -(S.V.I.) → ↓ coagulation)

3) 50% as free Ca⁺⁺ which can leave the plasma

i.e. total serum Ca⁺⁺ = Ca⁺⁺•albumin + Ca⁺⁺•cit/phos/sulfates + free Ca⁺⁺ ∴ ↓ in any → ↓ serum Ca⁺⁺ but only free Ca⁺⁺ is of critical importance

B.e.B. **HYPOCALCEMIA** [serum Ca⁺⁺ < 8.5 mg/dl]

B.e.B.a. **ETIOLOGIC FACTORS**

impaired ability to mobilize Ca⁺⁺ from bones

hypoparathyroidism (→ impaired mobility)

abnormal Ca⁺⁺ binding (too much in unionized form)

↑ pH, ↑ free fatty acids, rapid transfusion of citrated blood, acute pancreatitis

(NB, since Δ serum albumin → no Δ in free serum Ca⁺⁺, ↓ serum albumin rarely → manifestations)

abnormal losses

renal failure

decreased absorption from intestine

inactivity or vitamin D deficiency

B.e.B.b. **PATHOPHYSIOLOGY**

B.e.B.b.(A.) **ETIOLOGY OF HYPOCALCEMIA**

hyperventilation => respiratory alkalosis => / pH => / binding of Ca⁺⁺ to plasma protein } no Δ in total serum Ca⁺⁺ } / free Ca⁺⁺ < calcemia >

renal failure => / responsiveness of kidneys to PTH => / Ca⁺⁺ absorption from gut }

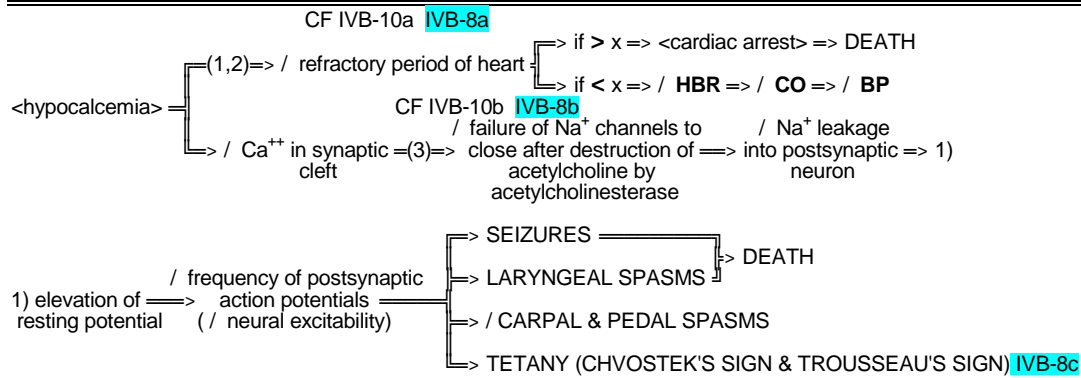
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Path IVB-9. General Etiology of Hypocalcemia

Key: 1: S IV.C: acid-base balance

2: F IVB-3: role of PTH on Ca⁺⁺ absorption from sm intestine

B.e.B.b.(B.) **MANIFESTATIONS OF HYPOCALCEMIA**



KEY: 1: Table IVB-1; 4e: Ca⁺⁺ maintains cardiac refractivity
 2: S VII.D.: cardiac function & dysfunction
 3: Table IVB-1; 4d: role of Ca⁺⁺ in **EPSPs & IPSPs**

IVB-8

Path IVB-10. General Manifestations of Hypocalcemia

B.e.C. **HYPERCALCEMIA** [total serum Ca⁺⁺ > 10.5 mg/dl]

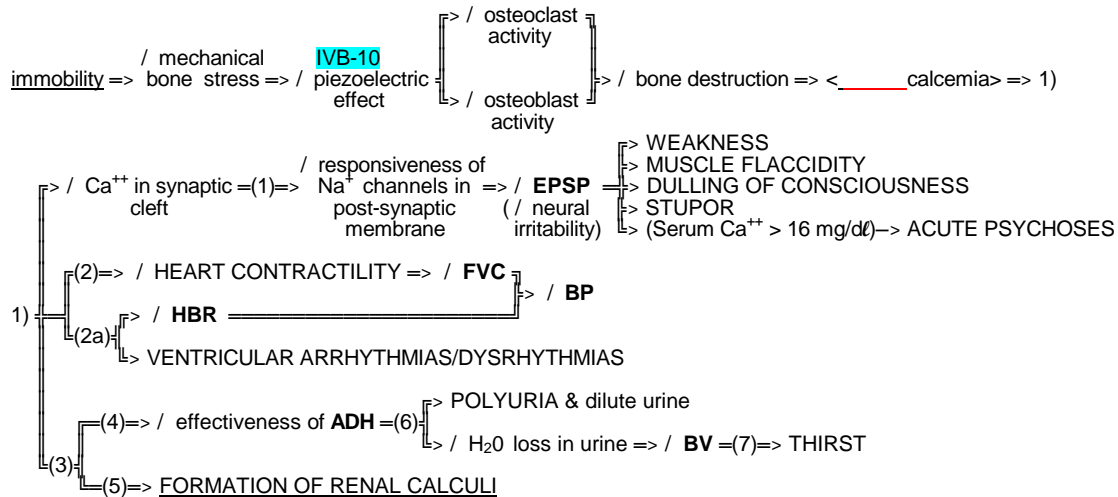
B.e.C.a. **ETIOLOGIC FACTORS**

- 1) excess osteoclast activity:
malignant neoplasms, ↑ bone destruction: immobility, thiazide diuretics, post menopausal, hyperparathyroidism.
- 2) **GI** tract: (sources: milk, egg yolk, shellfish, green-leafy vegies)
 ↑ intestinal absorption, excessive vitamin D, excessive Ca⁺⁺ in diet, milk-alkali syndrome

B.e.C.b. **PATHOPHYSIOLOGY**

manifestations results from:

- 1) ↓ neuromuscular activity
- 2) resorption of Ca⁺⁺ from bone
- 3) exposure of kidneys to high Ca⁺⁺ concentrations



KEY: 1: Path IVB-10: / Ca⁺⁺ in synaptic cleft necessary for Na⁺ channels to open
 2: Table IVB-1: 4a: binding of Ca⁺⁺ to troponin-tropomyosin complex *et seq.*; S VII.D: cardiac function & dysfunction
 2a: Table IVB-1; 4e: Ca⁺⁺ maintains cardiac refractivity
 3: S IX.A: renal function & dysfunction
 4: Ca⁺⁺ somehow interferes w/ activity of **ADH**
 5: F IVA-3: PO₄⁻³ excreted by kidney; S B.d.A: 3): if product of Ca⁺⁺ & PO₄⁻³ > 35, → precipitation of Ca₃(PO₄)₂
 6: CP IVA-3: **ADH** & urine volume
 7: Fig 16.8: p 429: ↓ **BV** & thirst

IVB-9 & HWA

Path IVB-11. Pathophysiology of Hypercalcemia